

New dinuclear Au(III) complex: Synthesis, characterization and study of the interactions with DNA

Novi dinuklearni Au(III) kompleks: Sinteza, karakterizacija i ispitivanje interakcija sa DNK

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Considering that gold(III) complexes are isostructural with platinum(II) complexes, the synthesis, characterization and study of their biological activity were very common in the last period.¹ We have synthesized new dinuclear gold(III) complex with 1.5-naphthyridine as bridging ligand. The structure of this complex was confirmed by different analytical methods (IR, UV-Vis, ¹H NMR, mass analysis, conductometry). The study of the interactions between complex and DNA was performed by UV-Vis spectrophotometry, fluorescence spectroscopy and viscosity measurement.



increasing concentrations of CT-DNA.



Table 1 The DNA-binding
constant (K_b) and Stern–Volmer
constant (K_{sv}) from EB–DNA
fluorescence competition
experiment. $K_b [M^{-1}] \times 10^5$ $K_{sv} [M^{-1}] \times 10^4$ 1.39 ± 0.02 3.939 ± 0.006

Fig. 5 Relative viscosity $(\eta/\eta_o)^{1/3}$ of CT-DNA (0.013mM) in buffer (0.01 M PBS, pH=7.4) in the presence of the increasing amounts of Au(III) complex (r).

CONCLUSION

Based on the results obtained by UV-Vis spectrophotometry can be concluded that complex binds to DNA. Additionally, according to the results obtained by fluorescence spectroscopy and by viscosity measurement, the covalent binding



Fig. 4 Emission spectra of EB bound to DNA in the presence of complex. [EB] = 6.73μ M, [DNA] = 6.73μ M; [Au] = $1.33 - 13.9 \mu$ M; $\lambda_{ex} = 527$ nm. The arrow shows the intensity changes upon increased concentration of complex.

References:

mode between complex and DNA was confirmed.

