

ABSTRACT BOOK

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LIST OF CONTENTS

Plenary lectures	s588
Parallel sessions	
Analytical chemistry Electrochemistry, Analysis, Sample manipulation	s593
Education and History, Professional chemists Ethics, Employability, Labels	s619
Environment and Green Chemistry	s635
European Young Chemists' Network	s667
Food Chemistry Food/Agriculture/Agrochemistry/Nanotechnology, food and processing	s671
Inorganic Chemistry plus Young inorganic chemistry day	s678
Life Sciences	s703
Nanochemistry/Nanotechnology/Molecular machines, Carbon tubes, sheets, balls	s729
Organic Chemistry, Polymers – I	s754
Organic Chemistry, Polymers – II	s780
Physical, Theoretical and Computational Chemistry	s807
Solid State Chemistry Materials chemistry/New materials	s832
Special Symposium: Jam Session plus Young National Winners in Bio-Organic Chemistry	s853
Poster session I	s859
Poster session II	s1109
List of Authors	s1368
List of Keywords	s1402

Poster session 2 - Organic chemistry

P-0840

KINETIC INVESTIGATION OF PHENYLSELENOETHERIFICATION OF SOME Δ^4 -ALKENOLS IN PRESENCE OF CATALYTIC AMOUNT OF CoCl_2 **V. DIVAC¹, M. KOSTIC¹, Z. BUGARCIC¹**¹ Faculty of Science, Department of Chemistry, Kragujevac, Serbia

The phenylselenoetherification of unsaturated alcohols is the one of the most rapid and convenient methods for formation of THF and THP type of rings. The convenient position of a double bond and a hydroxyl group in an alkenol can easily lead to formation of corresponding cyclization product. The reaction of selenium electrophiles with alkenols is a stereospecific *anti* addition where the nucleophile (pendant hydroxyl group) attacks usually at higher substituted carbon atom (Markownikoff stereoselectivity).¹ Despite the established synthetic utility of these selenofunctionalization reactions, there is no many report on a detailed kinetic investigation of these transformations.² In previous work it was found that in the presence and absence of various additives these reactions follow the mechanism of bimolecular nucleophile substitution $\text{S}_{\text{N}}2$ pathway.^{2c}

In this work we reported the kinetic study of the cyclization of some Δ^4 -alkenols with PhSeX ($\text{X}=\text{Cl}$, Br) in presence of catalytic amount of CoCl_2 . We investigated the influence of olefinic and carbinol substitution pattern in some Δ^4 -alkenols on the rate constants of the cyclization under the *pseudo*-first order conditions, in the presence and absence of CoCl_2 , by UV-VIS spectrophotometry. Reactions were carried out in the THF as a solvent. The obtained values for rate constants have shown that the reactions with phenylselenenyl bromide are slower then with chloride. Reactions with CoCl_2 present are faster then without one. Values for rate constants strongly depend on a substitution pattern of used alkenol.

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2-(METHYLAMINO)PYRIDINE AS A LIGAND PRECURSOR FOR THE TI-CATALYZED HYDROAMINOALKYLATION OF ALKENES AND STYRENES**J. DOERFLER¹, S. DOYE¹**¹ Institute of Pure and Applied Chemistry, Organic Chemistry, Oldenburg, Germany

Due to its direct and highly atom-efficient conversion of simple starting materials into more complex molecules by C–C bond formation the hydroaminoalkylation of alkenes^[1,2] and styrenes^[1c,2b] must be regarded as a very promising alternative to the industrial synthesis of amines. While reactions of 1-alkenes performed with group 5 metal catalysts (Ta, Nb) exclusively deliver the branched hydroaminoalkylation product^[1] mixtures of the branched and the linear product are obtained with Ti-catalysts.^[2] However, in this context, it must be noted that even with Ti-catalysts, the branched product is always formed as the major product and successful hydroaminoalkylations of styrenes can only be achieved with *N*-methyl anilines. Herein, we report the first intermolecular hydroaminoalkylation reactions of styrenes with *N*-alkyl anilines and dialkyl amines. The reactions are catalyzed by a 2-aminopyridinato Ti-complex^[3] that is generated in situ from $\text{Ti}(\text{NMe}_2)_4$ and the ligand precursor 2-(methylamino)pyridine. Most importantly, it is possible for the first time to obtain the industrially more interesting linear product as the major product of the reaction.

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