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Supporting information

Synthesis of Angularly Fused (Homo)triquinane Type Hydantoins as Precursors of Bicyclic α-Prolines

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			Compound		
	rac-(5R,6aS,9aS)- 3a	rac- (5S,6aS,9aS)- 3a	rac- (5R,6aS,9aS)- 3b	rac-(5R,6aS,9aS)- 3c	rac-(5S,6aS,9aS)-30
	$H(9)\alpha \leftrightarrow H(5)\alpha$	$H(9)\alpha \leftrightarrow CH_2^{Se}$	$H(8)\alpha \leftrightarrow H(5)\alpha$	$H(6a)\beta \leftrightarrow CH_2^{Se}$	$H(8)\alpha \leftrightarrow H(10)\alpha$
	H(6a)β↔H(8)β	$H(6)\alpha \leftrightarrow CH_2^{Se}$	$H(8)\alpha \leftrightarrow H(6)\alpha$	H(6a)β↔H(9)β	$H(8)\alpha \leftrightarrow H(6)\alpha$
NOE		H(6a)β↔H(8)β	$H(10)\alpha \leftrightarrow H(5)\alpha$	$CH_3 \leftrightarrow CH_2^{Se}$	$CH_3 \leftrightarrow CH_2^{Se}$
NUE		$H(6a)\beta \leftrightarrow H(5)\beta$	$H(6)\beta \leftrightarrow CH_2^{Se}$	$CH_3 \leftrightarrow H(10)\alpha$	$H(6)\beta \leftrightarrow CH_2^{Se}$
oneiations			H(6a)β↔CH ₂ ^{Se}	$H(6)\beta \leftrightarrow CH_2^{Se}$	H(6a)β↔H(9)β
				$H(8)\alpha \leftrightarrow H(10)\alpha$	$H(6a)β↔CH_3$
				$H(8)\alpha \leftrightarrow H(6)\alpha$	
	$CH_2^{Se} \leftrightarrow C(6)$	$CH_2^{Se} \leftrightarrow C(6)$	$CH_2^{Se} \leftrightarrow C(6)$	$CH_2^{Se} \leftrightarrow C(6)$	$CH_2^{Se} \leftrightarrow C(6)$
	$CH_2^{Se} \leftrightarrow C(5)$	$CH_2^{Se} \leftrightarrow C(5)$	$CH_2^{Se} \leftrightarrow C(5)$	$CH_2^{Se} \leftrightarrow C(5)$	$CH_2^{Se} \leftrightarrow C(5)$
	$H(5) \leftrightarrow CH_2^{Se}$	$H(5) \leftrightarrow CH_2^{Se}$	H(5)↔C(6)	$CH_2^{Se} \leftrightarrow CH_3$	$CH_2^{Se} \leftrightarrow CH_3$
	$H(5) \leftrightarrow C(3)$	H(5)↔C(6)	H(6)↔C(5)	$CH_3 \leftrightarrow C(6)$	$CH_3 \leftrightarrow CH_2^{Se}$
	H(6)↔C(9)	H(5)↔C(6a)	H(6)↔C(6a)	$CH_3 \leftrightarrow C(5)$	$CH_3 \leftrightarrow C(6)$
	H(6)↔C(9a)	H(5)↔C(9a)	H(6)↔CH ₂ ^{Se}	$CH_3 \leftrightarrow CH_2^{Se}$	$CH_3 \leftrightarrow C(5)$
	H(6)↔C(6a)	$H(5) \leftrightarrow C(3)$	H(6)↔C(10)	H(6)↔C(6a)	$H(6) \leftrightarrow C(5)$
	H(6a)↔C(1)	$H(6) \leftrightarrow C(7)$	H(6)↔C(10a)	$H(6) \leftrightarrow CH_3$	H(6)↔C(6a)
	H(6a)↔C(5)	$H(6) \leftrightarrow CH_2^{Se}$	H(6)↔C(10)	$H(6) \leftrightarrow C(7)$	$H(6) \leftrightarrow C(5)$
	H(6a)↔C(8)	H(6)↔C(6a)	H(6a)↔C(5)	$H(6) \leftrightarrow CH_2^{Se}$	H(6a)↔C(1)
	$H(7)\leftrightarrow C(8)$	$H(6) \leftrightarrow C(5)$	$H(6a) \leftrightarrow C(6)$	$H(6) \leftrightarrow C(5)$	$H(6a) \leftrightarrow C(5)$
	H(7)↔C(9)	H(6)↔C(9a)	$H(6a) \leftrightarrow CH_2^{Se}$	H(6)↔C(10a)	H(6a)↔C(6)
	H(7)↔C(6a)	H(6a)↔C(8)	H(6a)↔C(10a)	H(6a)↔C(8)	H(6a)↔C(7)
HMBC	H(7)↔C(9a)	$H(6a)\leftrightarrow C(1)$	H(7)↔C(8)	H(6a)↔C(7)	H(9)↔C(10a)
correlations	$H(7) \leftrightarrow C(1)$	$H(7) \leftrightarrow C(5)$	H(7)↔C(6a)	H(6a)↔C(6)	$H(10)\leftrightarrow C(9)$
	$H(8)\leftrightarrow C(1)$	$H(7) \leftrightarrow C(8)$	H(7)↔C(10a)	H(6a)↔C(10a)	H(10)↔C(10a)
	H(8)↔C(6a)	H(7)↔C(6)	H(7)↔C(1)	H(6a)↔C(1)	
	H(8)↔C(7)	H(7)↔C(6a)	H(8)↔C(9)	H(7)↔C(6)	
	H(8)↔C(9a)	H(7)↔C(9a)	H(9)↔C(8)	$H(7)\leftrightarrow C(8)$	
	$H(9) \leftrightarrow C(8)$	H(8)↔C(6)	H(9)↔C(10a)	$H(9) \leftrightarrow C(8)$	
	H(9)↔C(7)	$H(8) \leftrightarrow C(7)$	H(9)↔C(1)	H(9)↔C(10a)	
	H(9)↔C(6a)	$H(8) \leftrightarrow C(9)$	$H(10)\leftrightarrow C(1)$	H(9)↔C(1)	
	H(9)↔C(6)	$H(8) \leftrightarrow C(1)$	H(10)↔C(6a)	H(9)↔C(10)	
	H(9)↔C(9a)	$H(9) \leftrightarrow C(1)$	H(10)↔C(10a)	$H(10) \leftrightarrow C(1)$	
		H(9)↔C(8)		H(10)↔C(6a)	
		$H(9) \leftrightarrow C(7)$		H(10)↔C(10a)	
		$H(9) \leftrightarrow C(6a)$		$H(10) \leftrightarrow C(8)$	
		H(9)↔C(9a)			

 Table S1. Crucial Correlations Observed in NOESY and HMBC Spectra





























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f1 (ppm)



















-157.42





¹³C NMR (125 MHz, CDCl₃)

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~74.28 ~73.01

58.29 57.58





 ^{13}C NMR (50 MHz, D₂O + 2 drops of DCl 36%)

RP-230C RP-230 + DCl $<^{177.07}_{176.47}$






S71



S72

	(2c)	(3b)
Crystal data		
Chemical formula	$C_{12}H_{18}N_2O_2 \cdot H_2O$	$C_{17}H_{20}N_2O_2Se$
$M_{ m r}$	480.6	363.31
Crystal system, space group	Monoclinic, <i>I</i> 2/ <i>a</i>	Monoclinic, $P2_1/n$
Temperature (K)	294	294
a, b, c (Å)	12.7811 (4), 6.2674 (2), 33.3511 (9)	6.7697 (4), 18.4108 (9), 13.0431 (10)
β (°)	93.650 (3)	99.042 (7)
$V(\text{\AA}^3)$	2666.15 (14)	1605.43 (18)
Ζ	4	4
Radiation type	Μο Κα	Μο Κα
$\mu (\mathrm{mm}^{-1})$	0.09	2.35
Crystal size (mm)	0.49 imes 0.44 imes 0.04	$0.59 \times 0.15 \times 0.06$
Data collection		
Diffractometer	Gemini S (Oxford Diffraction)	Gemini S (Oxford Diffraction)
Absorption correction	Multi-scan	Multi-scan
T_{\min}, T_{\max}	0.852, 1	0.650, 1
No. of measured, independent and	11300, 3164, 2773	7207, 3665, 2432
observed $[I > 2\sigma(I)]$ reflections	0.010	0.000
$R_{\rm int}$	0.018	0.039
$(\sin \theta / \lambda)_{\rm max} ({\rm A}^{-1})$	0.685	0.683
Refinement		
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.067, 0.172, 1.14	0.066, 0.227, 1.07
No. of reflections	3164	3665
No. of parameters	167	203
No. of restraints	4	1
H-atom treatment	Mixture of independent and	Mixture of independent and
	constrained refinement	constrained refinement
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} ({\rm e} {\rm \AA}^{-3})$	0.35, -0.18	1.29, -0.93

Table S2 Crystallographic data, experimental and refinement details for 2c and 3b



Figure S1. Molecular structure of 2c. Ellipsoids are drawn at 40% probability level, whereas hydrogen atoms are depicted as spheres of arbitrary radii



Figure S2. Molecular structure of 3b. Ellipsoids are drawn at 40% probability level, whereas hydrogen atoms are depicted as spheres of arbitrary radii